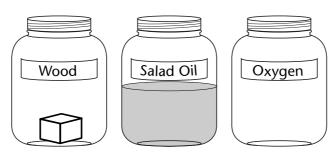
Solids, Liquids, and Gases • 3.1 Review and Reinforce

## **States of Matter**

## **Understanding Main Ideas**

Use the diagram to answer Questions 1 through 3. Write your answers on a separate sheet of paper.

- 1. Identify the physical state of the substances pictured below.
- **2.** What would happen to the shape of each substance if the jars were broken? Use the differences in the physical state of the substances to explain your answer.
- **3.** Would the volume of each substance change if each were moved into a larger container? Explain.



## **Building Vocabulary**

Match each term with its definition by writing the letter of the correct definition on the line next to the term.

<b>4.</b> solid	
<b>5.</b> liquid	
<b>6.</b> gas	
7. viscosity	
8. amorphous s	olid
9. crystalline so	lid
<b>10.</b> fluid	
11. surface tension	on

- **a.** a form of matter without a definite shape or volume
- **b.** the result of an inward pull among the molecules of a liquid that brings surface molecules closer together
- **c.** the resistance of a liquid to flowing
- **d.** a solid in which the particles are arranged in a regular, repeating pattern
- **e.** a form of matter that has a definite volume and a definite shape
- **f.** a substance that flows
- **g.** a form of matter that has a definite volume but no shape of its own
- **h.** a solid in which the particles are not arranged in any definite pattern